



# Ab Initio Protein Folding

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# From Sequence to Structure

- Does primary sequence lead to functional structure?
  - Isolate functional protein
  - Denature using urea or high temperature
  - Confirm loss of function
  - Reinstate folding conditions (remove urea, lower temp.)
  - Confirm gain of function
- In general protein sequence leads to functional structure
- Simulation of physical forces should allow computational folding of proteins
  - Levitt, M. and A. Warshel, *Computer simulation of protein folding. Nature, 1975. 253: p. 694-698.*



# Total Potential Energy

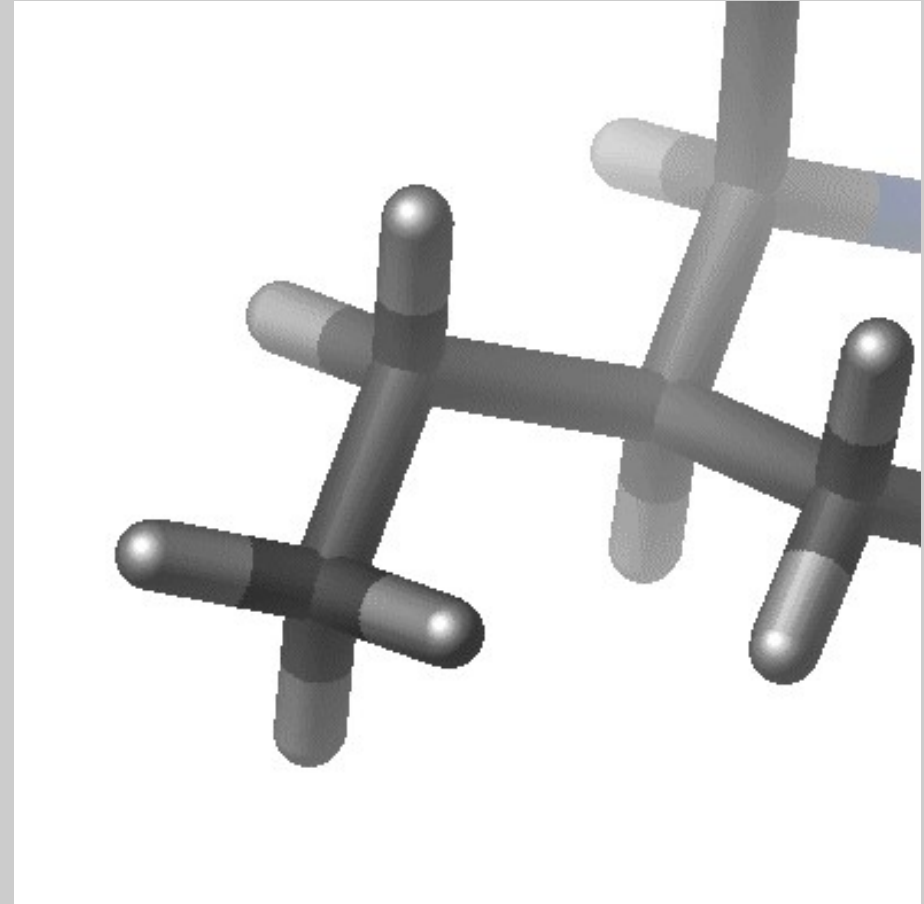
- Mathematical expression of the potential function is necessary for simulation of protein fold
- $E_{Total} = E_{Empirical} + E_{Effective}$ 
  - $E_{Empirical}$  : energy of the molecule as a function of the atomic coordinates
  - $E_{Effective}$  : restraining energy terms that use experimental information
- Neglect  $E_{Effective}$  term for true computational model
- Structure with the lowest total energy is the most stable structure
  - Justified by laws of Thermodynamics



# Potential Energy of Bond Lengths

- The bond length a pair of atoms is known empirically
- Bond lengths should not exceed the expected values
- Defined by two atoms

$$E_{Bond} = \sum_{bond} k_b (r - r_0)^2$$

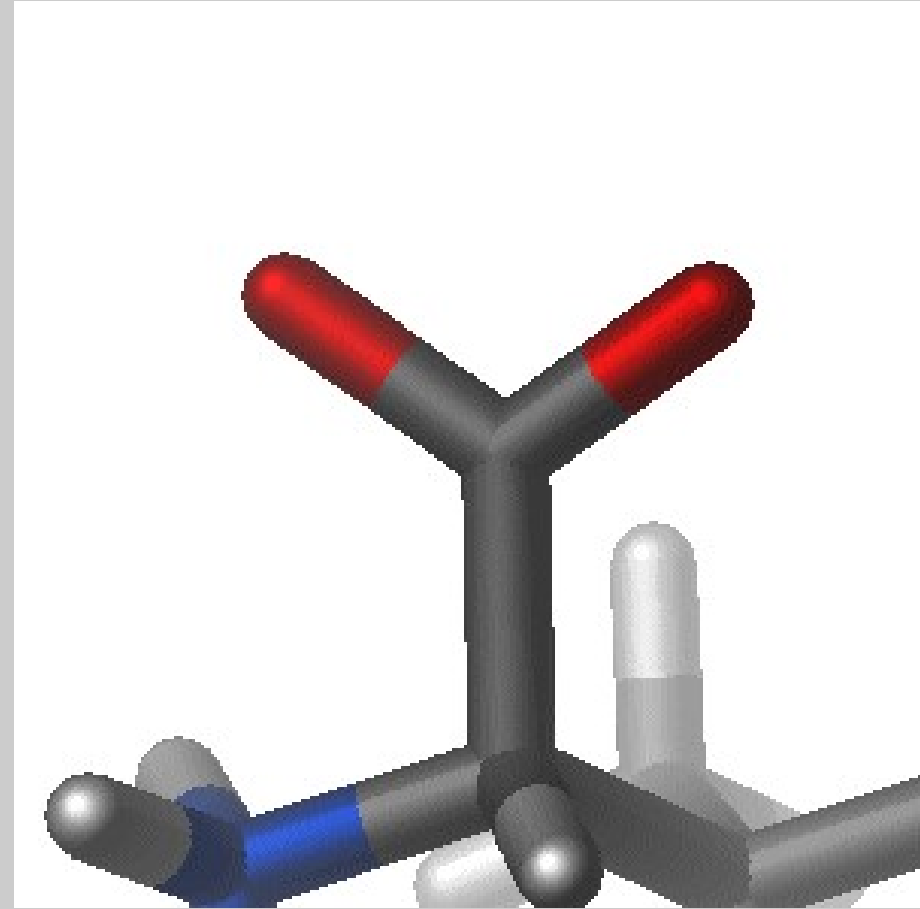




# Potential Energy of Bond Angles

- Bond angles should not deviate from the known quantities
- Involves three atoms

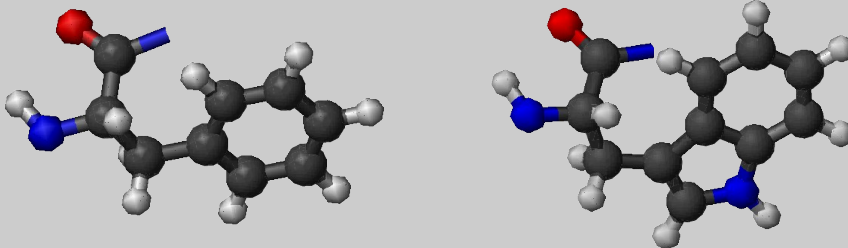
$$E_{Angles} = \sum_{angles} k_{\theta} (\theta - \theta_0)^2$$





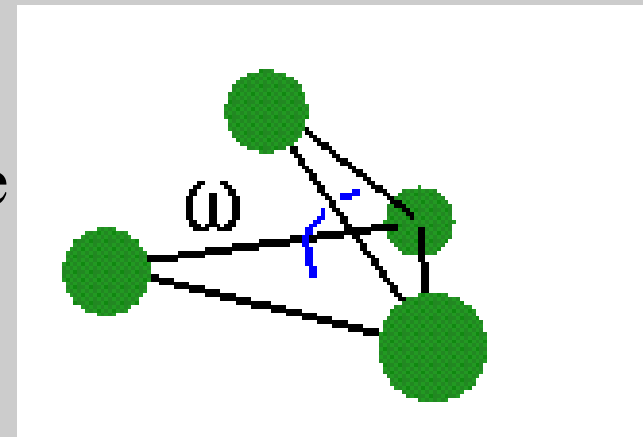
# P.E. of Improper Dihedrals

- Improper dihedrals represent the planarity of a group of atoms
  - Peptide plane
  - Aromatic side chains: phenylalanine, tryptophan, tyrosine, histidine



- Four atoms are required for this measure

$$E_{Impr} = \sum_{impr} k_i (\omega - \omega_0)^2$$





# Empirical Energy Terms

- All energies defined in terms of atomic coordinates of two, three or four atoms
- Conformational Energy Terms:
  - $E_{\text{BND}}$  : describes the covalent bond energy over all covalent bonds
  - $E_{\text{ANL}}$  : describes the bond angle energy over all bond angles
  - $E_{\text{DHE}}$  : describes the dihedral angle energy over all dihedrals
  - $E_{\text{IMR}}$  : describes the improper angle energies (planarity and chirality)
- Nonbonded Energy Terms:
  - $E_{\text{VDW}}$  : describes the energy of Van Der Waals terms
  - $E_{\text{HFC}}$  : describes the energy of electrostatic interactions



# Other Potential Terms

- Hydrophobic and hydrophilic interaction.
  - Requires presence of water in the simulation.
  - Addition of water to the simulation is difficult.
  - Will require identification of cavities and calculation of movement of water molecules.
- Hydrogen bonds:
  - Also requires assessment of water accessibility.
  - Water interferes with formation of hydrogen bonds.
- Gas phase simulation
  - Absence of water.
  - Computationally much more convenient



# Total Energy Term

- $E_{\text{Total}}$  is the total potential energy of a conformation
- Force Field: A vector field representing the gradient of the total potential
- $\omega$  is referred to as the force constants

$$E_{\text{Total}} = \sum \left[ w_{\text{BOND}}^p E_{\text{BOND}} + w_{\text{ANGL}}^p E_{\text{ANGL}} + w_{\text{DIHE}}^p E_{\text{DIHE}} + w_{\text{IMPR}}^p E_{\text{IMPR}} + w_{\text{VDW}}^p E_{\text{VDW}} + w_{\text{ELEC}}^p E_{\text{ELEC}} \right]$$

$$E_{\text{BOND}} = \sum_{\text{bonds}} k_b (r - r_0)^2$$

$$E_{\text{ANGL}} = \sum_{\text{angles}} k_\theta (\theta - \theta_0)^2$$

$$E_{\text{DIHE}} = \sum_{\text{dihedrals}} \sum_{i=1, m} \begin{cases} k_{\varphi_i} (1 + \cos(n\varphi_i + \delta_i)) & n_i > 0 \\ k_{\varphi_i} (\varphi_i - \delta_i)^2 & n_i = 0 \end{cases}$$

$$E_{\text{IMPR}} = \sum_{\text{imprers}} \sum_{i=1, m} \begin{cases} k_{\varphi_i} (1 + \cos(n\varphi_i + \delta_i)) & n_i > 0 \\ k_{\varphi_i} (\varphi_i - \delta_i)^2 & n_i = 0 \end{cases}$$

$$E_{\text{VDW}} = \sum_{\text{vdw}} \frac{A_{ij}}{R_{ij}^{12}} - \frac{B_{ij}}{R_{ij}^6}$$

$$E_{\text{ELEC}} = \sum_{i, j} \frac{q_i q_j}{4 \pi \epsilon_0 r_{ij}}$$



# Force Field

- Technically, the derivate of the potential energy
  - A vector field of forces
- Some currently existing force fields (forcefield):
  - Xplor-NIH
  - AMBER
  - CHARMM
  - MM2, MM3 and MM4
  - Sybyl
  - Etc.



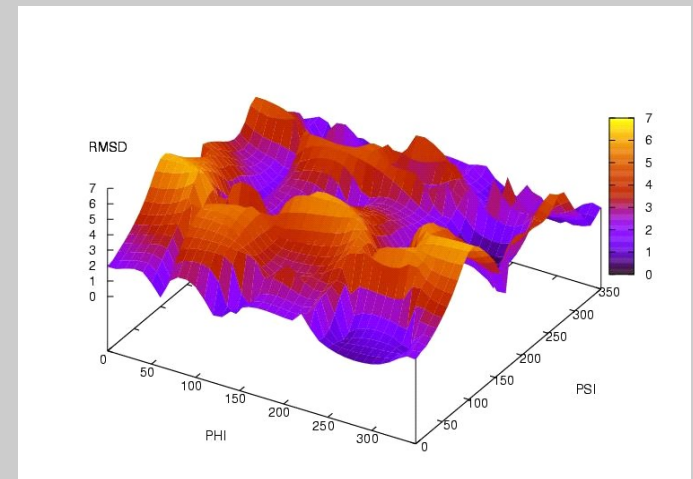
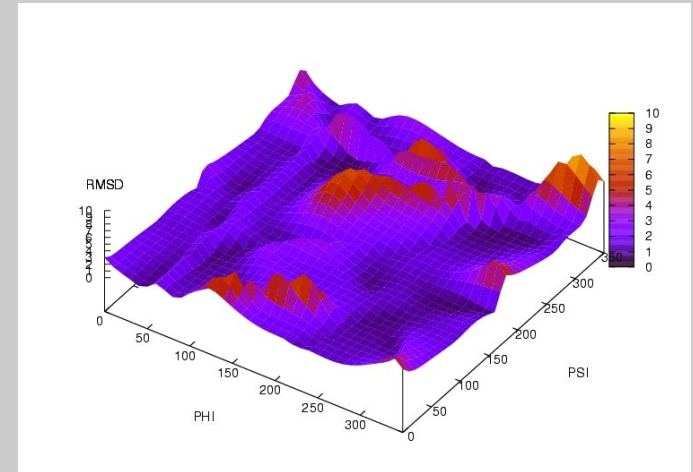
# Minimization of Total Energy

- Theoretically, the structure with the minimum total energy is the structure of interest.
- A number of minimization algorithms can be utilized.
  - Gradient descent
  - Monte Carlo and Simulated Annealing
  - Newton's
  - Genetic Algorithm
  - Distributed Global Optimization
  - Branch and Bound
  - ...



# Complexity of The Problem

- Assuming a protein with 100 residues and in average 10 atoms per residue, what is the complexity of this problem?
- What are the variables of this problem? How many?
- How complex is the total energy landscape?
- How costly is each evaluation of the  $E_{\text{Tot}}$  and its gradient?
- Beyond our computational capabilities.





# Computational Complexity of Protein Folding

- For a protein of size  $N$  amino acids:
  - $df = 2 \times (N - 1)$
  - Each degree of freedom spans  $0^\circ$ - $360^\circ$
  - Possible conformations at  $10^\circ$  resolution:  $36^{2(N-1)}$
  - $N = 100$ ,  $10^6$  struct / sec  $\Rightarrow$  4.4575E+291 millennia
  - NP class of problems.
  - $N=11 \Rightarrow$  32 millennia
  - $N=11$ , 50 angles  $\Rightarrow$  1 millennium

